Computer Simulations of Alzheimer's Amyloid β -Protein Folding and Assembly

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Abstract: Pathological folding and aggregation of the amyloid β -protein (A β) are widely perceived as central to understanding Alzheimer's disease (AD) at the molecular level. Experimental approaches to study A β self-assembly are limited, because most relevant aggregates are quasi-stable and inhomogeneous. In contrast, simulations can provide significant insights into the problem, including specific sites in the molecule that would be attractive for drug targeting and details of the assembly pathways leading to the production of toxic assemblies. Here we review computer simulation approaches to understanding the structural biology of A β . We discuss the ways in which these simulations help guide experimental work, and in turn, how experimental results guide the development of theoretical and simulation approaches that may be of general utility in understanding pathologic protein folding and assembly.

Keywords: Alzheimer's disease, amyloid β -protein, computer simulations, coarse-grained protein model, protein folding, oligomer formation, fibril formation.

I. INTRODUCTION

Searching for the primary causative event leading to neuronal injury and loss in Alzheimer's disease (AD) has been at the heart of AD research for the past century. Alois Alzheimer first defined the clinicopathological syndrome that bears his name at a meeting in Munich in 1906. Use of electron microscopy in the 1960's allowed Michael Kidd and Robert Terry to describe the ultrastructure of two types of lesions in the cerebral cortex of Alzheimer patients, senile plaques and neurofibrillary tangles [1,2]. However, it was only in 1984, when Glenner and Wong isolated and sequenced amyloid β-protein (Aβ) from cerebrovascular deposits, that plaques were linked to accumulation of A β [3]. It has been postulated that pathological aggregates of A β are the cause of all forms of AD, whereas the appearance of neurofibrillary tangles and other neuropathological changes are a consequence of pathological A β assembly [4]. If this postulate, also known as the amyloid cascade hypothesis, is correct, then genetic alterations that produce AD should be related to $A\beta$ aggregation. Studies of familial forms of AD indeed show that naturally-occurring mutations strongly influence A β metabolism and aggregation [5-7]. In addition, the amyloid hypothesis has received strong support by experiments using transgenic animal models of AD [8-11].

A β is a normally-secreted, soluble proteolytic product of the amyloid β -protein precursor (APP). APP is a transmembrane protein that undergoes sequential endoproteolytic cleavage to produce predominately two forms of A β , A β (1-40) and A β (1-42). Although A β (1-40) is the major species produced, the principal species deposited within the parenchyma of the AD brain is A β (1-42) [12,13]. Senile plaques, which are believed to represent the final product of A β aggregation, contain A β fibrils. A β fibrils have specific tinctorial properties, including birefringence when treated with the amyloidophilic dye Congo Red and visualized using crosspolarized light. Fibrils can be observed in the electron microscope, which reveals filaments ranging from 60 Å to 130 Å in width [14], and characterized by powder X-ray diffraction, which produces patterns consistent with a cross- β core structure [15].

In vitro studies have shown that AB fibril assembly involves the formation of a number of intermediates, including oligomers and protofibrils (for a recent review, see Ref. [16]). A precise determination of the precursor-product relationships among the many different types of AB assemblies that have been described is lacking. There may be more than one pathway of assembly. For example, in addition to senile plaques that are associated with the fibrillar form of A β , there exist diffuse plaques with amorphous A β aggregates and these are far more abundant in the AD brain (for a review, see Ref. [17]). According to a traditional view, after $A\beta$ is generated from APP, it is secreted in a monomeric form into the extracellular space where it forms aggregates that eventually form senile plaques. On the other hand, there is evidence that some $A\beta$ is generated intracellularly and that the intracellular A β could play a pathophysiologic role in AD and Downs syndrome (for a review, see Ref. [18]).

In vitro studies demonstrate that synthetic A β monomers aggregate into A β oligomers that eventually undergo structural changes and assemble into protofibrils and fibrils [19-22]. During the last decade, *in vitro* and *in vivo* evidence has implicated the soluble A β oligomers as the predominant neurotoxic species (for reviews, see Refs. [23,24,25]). Even at

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nanomolar concentrations, $A\beta$ oligomers were shown to kill neurons in hippocampal slices [26] and inhibit long-term potentiation *in vivo*, which might lead to impairment of learning and memory in AD [27,28]. In addition, $A\beta$ oligomers were found to disrupt the membrane and cause calcium dysregulation in cells in culture, which might lead to cell death [29]. Human $A\beta$ oligomers injected into rats were shown to directly interfere with cognitive function without inducing permanent neurological deficits [30]. Recently, a new and highly stable oligomer species made of 12 $A\beta$ (1-42) peptides, named $A\beta$ (1-42) globulomers, that also exists in the AD brain, was prepared *in vitro* and shown to block long-term potentiation in rat hippocampal slices [31].

Detailed, quantitative analyses of the three-dimensional structure, energetics, and dynamics of oligomer formation are necessary steps toward a molecular understanding of $A\beta$ assembly neurotoxicity. During the formation of fibrils, oligomers of different sizes co-exist with monomers and larger aggregates such as protofibrils and fibrils. The relative amounts of each oligomer type are small. Although important information can be obtained by using solid state NMR [32], quasielastic light scattering [33], and other methods, *in vitro* structure determination of oligomers is limited. Limitations also are caused by the metastability and short lifetimes of oligomers. Computer simulations, in contrast, are not subject to the same kinds of problems, allowing small oligomers to be studied at full atomic resolution.

During the past two decades, computational power has increased, allowing longer, all-atom molecular dynamics simulations, development of efficient new methods for sampling configurational space, and optimization of force-field parameterization to enable study of explicitly-solvated proteins on time scales of up to several hundreds of nanoseconds (for reviews, see Refs. [34,35]). However, to examine biologically-relevant dynamics and interactions, such as folding, protein-protein docking, rearrangement upon ligand binding, and protein assembly, time scales of microseconds and longer are needed. To achieve such long simulation times, simplified representations of proteins have been employed, a strategy proposed three decades ago [36]. In the past decade, a variety of coarse-grained protein models, which integrate a large number of degrees of freedom into a few, have been developed and used extensively (for reviews, see Refs. [37,38,39]).

In this review, we classify the computational approaches and present up-to-date simulation studies of A β folding and assembly. The simulation studies are organized into two major sections, the first one on A β monomer folding and the second one on A β assembly. Within each of these two sections, studies on smaller fragments are presented first, followed by larger fragments, and finally full-length A β (1-40) and A β (1-42). To ensure biological relevance of the computational studies, most were based on prior experimental approaches and findings. For this reason, we also present results of selected experimental studies.

II. CLASSIFICATION OF COMPUTATIONAL APP-ROACHES

Computational methods applied to study amyloid aggregation *in silico* can be characterized by: (a) the type of the method used (molecular dynamics, discrete molecular dynamics, Monte Carlo, Langevin dynamics, Brownian dynamics, activation-relaxation technique);

(b) the way the solvent is treated (implicitly or explicitly);

(c) the technique used to sample the configurational space (umbrella sampling, replica-exchange, weighted histogram analysis method); and

(d) the degree of complexity of amino acid description (one-bead, two-bead, four-bead, six-bead, united-atom models as described below).

Molecular dynamics (MD) and discrete molecular dynamics (DMD) are the most commonly used methods to study protein folding and aggregation. To emphasize that in MD the protein is usually modeled by all atoms, the term "all-atom MD" is sometimes used. In DMD a simplified description of the protein is mostly used: each amino acid can be represented by only one "atom" or "bead" (one-bead model), by one backbone and one side chain beads (twobead model), by three backbone and one side chain beads (four-bead model), by four backbone and up to two side chain beads (six-bead model) or by all atoms except hydrogens (united-atom model). Even though MD and DMD are the most commonly used method, other methods, such as Monte Carlo, Langevin, and Brownian dynamics, typically associated with a simplified description of a protein, have been applied to study $A\beta$ folding and aggregation. Monte Carlo techniques are mostly used to obtain global knowledge of the phase space, including the intermediate and denatured states of proteins [40]. Because folding is a slow process compared to microscopic motion of a protein chain, overdamped Langevin dynamics is applied to the study of proteins. The Langevin equations of motion include the viscosity of the solvent as well as a white Gaussian random force with zero mean derived from the fluctuation-dissipation theorem that models thermal fluctuations at a given temperature [41,42,43,44]. In Brownian dynamics, each particle moves in a force field given by the interparticle potential with individual diffusion coefficient, while thermal fluctuations are accounted for in a similar way as in the Langevin dynamics [45].

Simulations of complex systems such as biopolymers at a constant volume and temperature are subject to a multipleminima problem, because systems simulated at low temperatures tend to get trapped in one of multiple local-minimum energy states (for a review, see Ref. [46]). To overcome this problem, several techniques have been proposed, such as umbrella sampling [47,48,49,50], replica-exchange [51], simulated tempering [52], and weighted histogram analysis methods [53]. A simulation in such a generalized ensemble realizes a random walk in the potential energy space, alleviating the multiple-minima problem. These techniques can be used in combination with MD, DMD, Monte Carlo, or Langevin dynamics.

The method with the most detailed information is the allatom MD with explicit solvent, where the solvent molecules are explicitly modeled. Many different force fields are available that differ in the way electrostatic interactions are implemented and in the model for water (SPC, TIP3P, TIP4P, TIP5P). In MD of proteins in explicit solvent, the simulation time is limited to several hundred nanoseconds. Longer times are accessible using all-atom MD with implicit solvent [54], where the absence of solvent molecules drastically reduces the number of particles in the simulation. Recently, a replicaexchange MD approach [51] with implicit solvent has been applied to study dimer formation of amyloidogenic peptides [55]. Within the replica-exchange technique, simulations of a number of copies (replicas) of the system under study are conducted in parallel. Temperature is a dynamic variable and is individually assigned to each replica. Periodically, after a fixed number of simulation steps, two replicas are swapped with a probability that depends on the potential energy and the temperature difference. This technique allows for unrestrained walks for each replica up and down in temperature between two limiting temperatures. Consequently, the dynamical processes are accelerated due to crossing of the potential energy barriers that would not be possible in constant temperature simulations.

The protein aggregation process occurs over time scales larger than a microsecond, so all-atom MD with explicit solvent is not usable. The activation-relaxation technique (ART) was recently introduced using the implicit solvent incorporated into the interaction potential [56,57,58] (for a review, see Ref. [59]). ART consists of three general steps: (1) the first activation step during which a random deformation is applied to "kick" the initial conformation out of the local minimum; (2) the second activation step during which the conformation is pushed in the direction of the lowest negative curvature on the energy landscape, while the energy is minimized; and (3) the relaxation step during which the energy is minimized until the total force approaches zero. The advantage of this method is that the system moves through the conformational space rapidly without need to wait for a rare event to occur. However, because the entropy is not taken into account, temperature is not defined by this technique. In addition, the dynamics of the system cannot be directly related to the observed pathways of protein folding and assembly.

The idea of applying a fast and efficient discrete molecular dynamics method (DMD) [60] to study protein folding was proposed in 1996 [61]. Soon after, the method was applied to study protein folding and assembly [62,63,64,65,66]. The main simplification in this method is to replace continuous interparticle potentials by a square-well or a combination of square-well potentials. Consequently, particles move along straight lines with constant velocities until a pair of particles reaches a distance at which the interparticle potential is discontinuous. A collision event then takes place during which the velocities and directions of the particles are updated while preserving the total kinetic energy, momenta, and angular momenta. Because DMD is event-driven, it is faster than all-atom MD. Even though it is not a requisite, DMD is typically used without explicit solvent. Solvent effects are implemented through effective interparticle potentials. In addition, protein models are usually coarse-grained to further increase the efficiency. Among the coarse-grained models, the four-bead [67,68,69,70], six-bead [71], and united-atom models [72] are particularly important as they allow for an *ab initio* simulation approach [73].

III. $A\beta$ FOLDING

The primary structure of $A\beta(1-42)$ is defined by the amino acid sequence: DAEFRHDSGYEVHHQKLVFFAED VGSNKGAIIGLMVGGVVIA. The last two amino acids I41 and A42 are missing in the alloform $A\beta(1-40)$. At physiological conditions and neutral pH, amino acids D, E, K, and R are charged, while H is typically considered neutral. Most hydrophobic amino acids are concentrated in the central hydrophobic core (L17-A21) and in the C-terminal part of the peptide (A30-A42), where with exception of G33, G37, and G38, all amino acids are hydrophobic. The first 15 N-terminal residues are predominantly hydrophilic.

In this section, we describe first the experimental findings on folding of specific A β peptides, and review the relevant computer simulation work. The computer simulation studies encompass folding of a decapeptide, A β (21-30), folding of a longer fragment, A β (10-35), and its Dutch mutant, and folding of full-length A β (1-40) and A β (1-42).

A. Folding Events of $A\beta(21-30)$

Lazo et al. used limited proteolysis with mass spectrometry to identify protease-resistant segments of $A\beta(1-40)$ and A β (1-42) [74]. They showed that a ten-residue segment, A21-A30, was protease resistant in both alloforms. A similar protease resistance was observed with the A β (21-30) decapeptide alone. Solution-state NMR studies of the decapeptide revealed a turn formed by the peptide segment V24-K28. Lazo et al. postulated that this turn nucleates the folding of a full-length A β monomer, which represents the first step in A β assembly. They also proposed three factors contributing to the formation and stability of the turn: (1) the innate propensity of G25, S26, and N27 to exist in turns; (2) the salt bridges E22-K28 or D23-K28; and (3) hydrophobic interactions between the isopropyl side chain of V24 and the *n*-butyl portion of the K28 side chain. Comparing the protease-resistant regions of A β (1-40) and A β (1-42), they also found that the fragment V36-A42 was protease resistant in A β (1-42) only. This result is consistent with previous experimental studies suggesting a β -helix model of A β (34-42) fibrils with a turn at G37-G38 stabilized by hydrophobic interactions [75], leading to the observed loop or turn-like conformation in the G37-G38 region of $A\beta(34-42)$ [76].

Borreguero et al. used DMD simulations and a unitedatom protein model with implicit solvent to study folding events of A β (21-30) [72]. All heavy atoms (all atoms except hydrogens) were modeled and the amino acid-specific interactions between pairs of side chain atoms were derived using an experimental estimation of the gain/loss of the free energy on transferring a particular amino acid from an aqueous solution to a gas phase [77]. The results of this study showed that A β (21-30) folds into a loop-like conformation driven by the hydrophobic attraction between the side chains of V24 and K28. Comparing the simulation results to the in vitro study [74], Borreguero et al. found that transient D23-K28 and E22-K28 salt bridges stabilized the loop conformation, with the E22-K28 interaction being more favorable. They also showed that at high strengths of the electrostatic interaction, which occur in desolvated regions inside proteins and aggregates, A β (21-30) folding is destabilized by the D23K28 salt bridge interaction, in agreement with studies of molecular modeling of full-length A β [78].

Recently, Cruz *et al.* studied the dynamics of A β (21-30) and its E22Q ("Dutch") mutant using all-atom MD and explicit water, salted water, and reduced-density water [79]. The folding events due to hydrophobic interaction between V24 and K28 side chains were observed in the wild type peptide in water at physiological conditions, in agreement with Borreguero et al. [72]. In contrast, the Dutch mutant did not display any stable conformation events. In salted water, the salt bridges E22-K28 and D23-K28 played an important role in stabilizing the A β (21-30) folding events, while in reduced-density water stable α -helix conformations were observed. These results indicate that the folding of $A\beta(21-$ 30) decapeptide, and possibly of full-length A β , is strongly sensitive to small solvent changes. If folding of the fragment A21-A30 nucleates the folding of full-length A β , as suggested by Lazo et al. [74], then subtle changes in the solvent environment could disrupt the folded structure and consequently change the pathway of A β assembly.

Conformational space accessible to the A β (21-30) fragment was investigated by Baumketner *et al.* using replica exchange MD in explicit solvent [80]. Baumketner *et al.* found conformations belonging to the global free energy minimum in good agreement with reported NMR structures [74].

B. Folding and Solubility of $A\beta(10-35)$

NMR experiments on A β (10-35) in an aqueous solution revealed a collapsed structure with loops, strands, and turns without any significant amount of α -helical or β -strand content [81]. This is in contrast to the full-length A β monomer structures in a membrane-like environment that show two α helical regions separated by a kink at V24-G29 [82,83].

Massi et al. performed all-atom MD simulations of A β (10-35) monomer in an aqueous environment at the nanosecond time scale [84]. The starting structures for the simulations were derived from distance geometry calculations employing NMR-derived constraints. Although their simulations indicated that the peptide is somewhat disordered in solution, the central hydrophobic core, L17-A21, and the turn region, V24-N27, were particularly stable. The results indicated that these two structural motifs are cooperatively stabilized through intramotif hydrogen bonds. Massi et al. expanded their study by performing all-atom MD of the A β (10-35) monomer in an aqueous environment in both the wild type and the Dutch mutant forms [85]. They generated and analyzed multiple nanosecond time scale MD trajectories using measures of the peptide's average structure, hydration, conformational fluctuations, and dynamics. The results showed that the Dutch mutant, as well as the wild type peptide, are stable in collapsed coil conformations consistent with the experimentally-derived structure of Zhang et al. [81]. However, the Dutch mutant is more open and flexible and the E22Q substitution leads to a change in the first solvation layer of water molecules close to the peptide's hydrophobic patch that results in increased solvation. Massi et al. argued that these results may explain the increased propensity for fibril deposition in the Dutch mutant.

An increased fibril elongation rate of the Dutch mutant of the A β relative to the wild type peptide was reported by several experimental groups [86,87]. Massi *et al.* disproved the hypothesis that the increased fibril elongation rate is due to a larger propensity for the formation of β -structure in the Dutch mutant monomer relative to the wild type peptide [88]. They applied all-atom MD simulations to study the wild type and the Dutch mutant forms of A β (10-35) in an aqueous solution at a four nanosecond time scale. The simulation results indicated that the propensity for formation of a β -structure is no greater in the E22Q mutant peptide than in the wild type peptide. Massi *et al.* proposed that the greater stability of the solvated wild type peptide relative to the Dutch mutant leads to a decreased fibril elongation rate in the latter. They concluded that this stability difference is due

to the differing charge state of the two peptides.

In a subsequent study, Massi and Straub performed allatom MD simulations to analyze the structure, energetics, and dynamics of water in the first solvation shell of the A β (10-35) monomer and the Dutch mutant peptide [89]. They analyzed the hydrogen bond network, the water residence times for each residue in the peptide, the diffusion constant, experimentally-determined amide proton exchange, and the transition probabilities for water to move from one residue to another or into the bulk. The results indicated that in both peptides the water molecules at the peptide-solvent interface are organized in an ordered structure different from that of the bulk water. In addition, they showed that the peptide structure inhibits water diffusion perpendicular to the peptide surface by a factor of 3 to 5 relative to diffusion parallel to the peptide surface, which is comparable to the diffusion of the bulk water. Water in the first solvation shell showed dynamical relaxation on fast (~1-2 ps) and slow (~10-40 ps) time scales, indicating intricate water dynamics that may influence the peptide dynamics at different levels. The interaction between the peptide and the solvent was found to be stronger in the wild type than in the Dutch mutant peptide, suggesting that the wild type is characterized by a more stable folded structure than the Dutch mutant.

Straub et al. introduced the MaxFlux algorithm for the computation of likely pathways of global macromolecular conformational transitions [90]. The algorithm is capable of describing intermediate structural states between two known (initial and final) structural states, thus rendering a pathway. The algorithm assumes an overdamped diffusive dynamics for the biomolecule. As an application of the MaxFlux method, Straub et al. explored conformational transitions between α -helical, collapsed coil, and β -sheet conformations of the A β (10-35) monomer generating possible intermediate structures, i.e., pathways, between the two conformations. The results showed that there are significant energy barriers between the collapsed coil and α -helical conformations and between the collapsed coil and β -sheet conformations. Each transition pathway involves an early formation of the turn in the region V24-N27, which has well-preserved structure in both α -helical and collapsed coil conformations. This result suggested that by destabilizing this turn, the kinetics of α helix and collapsed coil formation could be altered.

Recently, Han and Wu studied conformations of $A\beta(10-35)$ peptide using all-atom MD with explicit water sol-

vent [91]. The results of this study revealed that the collapsed coil structure determined by experiments is stable at pH 5.6 for hundreds of nanoseconds, but it exchanges with a strand-loop-strand conformation on the millisecond time scale. The strand-loop-strand conformation is characterized by a loop at residues D23-K28 which allows the central hydrophobic core (L17-A21) to be in hydrophobic contact with the residues A30-M35. Because the strand-loop-strand conformation resembles an intramolecular arrangement of the $A\beta(10-35)$ peptide in a fibril, Han and Wu suggested that this conformation represents an important intermediate on the pathway from monomer folding to fibril formation. They also performed simulations at pH 2.0, to mimic the mutations E22Q and D23N, and showed that at a temperature of 400 K the strand-loop-strand conformation is considerably populated, while the collapsed coil structure is disrupted. These results are in agreement with recent finding by Lazo et al. on folded A β (21-30) monomer conformation with a loop centered at G25-S26 [74] and give support to the hypothesis that the strand-loop-strand conformation plays a role of an intermediate structure on the pathway to fibril formation.

C. Folding of Full-Length Aβ

Experimental studies of $A\beta(1-40)$ and $A\beta(1-42)$ monomers in water–organic solvent mixtures showed that the monomer structure of both consisted of two α -helical regions connected through a flexible turn- or bend-like kink [82,83]. The temperature-dependence of the $A\beta(1-40)$ monomer and dimer structures in water was experimentally studied by Gursky and Aleshkov [92]. They found a folded $A\beta(1-40)$ monomer structure with no substantial amounts of α -helix or β -strand at low temperatures. As the temperature was increased to physiological, a substantial β -sheet content was detected. This structural transition was not accompanied by oligomer formation, thus it was attributed to $A\beta(1-40)$ monomers and dimers. These studies showed that full-length $A\beta$ monomer structure is very sensitive to external conditions, such as temperature, pH, and solvent.

Typically, all-atom MD studies are limited to ~100 ns time scales, which is not sufficient to span the entire folding process even in the case of a relatively short protein, such as full-length A β . Thus, numerous studies have taken simplified approaches by applying coarse-grained protein models. Using a two-bead amino acid model with one backbone and one amino acid-specific side chain bead and Brownian dynamics, Mukherjee and Bagchi showed that $A\beta(1-40)$ folds into a conformation characterized by a β-bend at residues V24-K28 [45]. These simulations also showed that $A\beta(1-40)$ folding dynamics consists of several stages: a two-stage fast hydrophobic collapse and a slow decay stage during which the protein formed more favorable pair contacts to replace the less favorable ones [45]. Despite the lack of a unique native state, the multistage dynamics of A β (1-40) folding is in accord with a predicted protein folding dynamics in which a slow, rate-determining step is related to a search through the conformational space to find a transition state from which the protein folds rapidly to the lowest-energy state [93].

A DMD study by Urbanc *et al.* using a four-bead amino acid model with hydrogen bonds and amino acid-specific interactions not only addressed the formation of $A\beta(1-40)$

and A β (1-42) oligomers, but also studied monomer folding prior to oligomer formation [94]. Results of this study showed that a folded A β (1-42) monomer, but not A β (1-40) monomer, possesses a turn at G37-G38 stabilized by a hydrophobic interaction between V36 and V39 [94]. This turn structure is in agreement with the solution ¹H NMR study of the C-terminal fragment A β (34-42) by Weinreb *et al.* who suggested that this C-terminal hydrophobic cluster nucleates amyloid formation in AD [95]. Moreover, Urbanc *et al.* showed that this turn is present in oligomers of both A β (1-40) and A β (1-42), implicating a possible important role of this turn in oligomer formation.

One of the most extensive all-atom MD studies of helixto-coil conformational change in A β (1-40) monomer was reported by Xu *et al.*, who studied A β (1-40) folding in both aqueous and membrane-like environments [96]. In an aqueous solution, A β (1-40) trajectories showed an α -helix $\rightarrow \beta$ sheet and a β -sheet \rightarrow random coil conformational change. The residues V24-G37 represented the core of the β -sheet conformation that would presumably be important for fibril formation, while G25, G29, G33, and G37 were shown to be essential for formation of the β -sheet structure. Xu et al. showed that in a biomembrane environment, $A\beta(1-40)$ prefers an α -helix conformation, however the protein has a tendency to exit the membrane environment. In another all-atom MD study of A β (1-42) folding in an aqueous solvent at various temperatures and pH conditions, Flöck et al. showed that at least one of the two α -helices is not stable, but rather rapidly converts to a random and β -strand rich conformation [97]. Both studies are in qualitative agreement with the experimental findings.

Recently, an A β (1-42) monomer structure was studied by Baumketner *et al.* using a combination of ion-mobility mass spectrometry and replica exchange MD simulations with implicit water solvent [98]. They showed that $A\beta(1-42)$ did not adopt a unique fold, but rather a mixture of rapidly interconverting conformations that were classified into three distinct families. The secondary structure analysis revealed that these conformations were dominated by the loops and turns, but also showed some helical structure in the C-terminal hydrophobic tail. Baumketner et al. proposed that an increase in α -helical structure as observed in A β intermediates by Kirkitadze et al. [99] results from association of unstructured monomers into oligomers in such a way that the hydrophobic tails of the peptide become shielded from the solvent. This shielding would create an apolar microenvironment promoting α -helical structure to grow from preexisting seeds. This hypothesis is amenable to both in silico and in vitro testing.

IV. Aβ AGGREGATION

Understanding pathways of oligomer and fibril formation of full-length A β as well as discerning the structures of folded monomers, oligomers, and fibrils of full-length A β is the aim of current computer simulation studies. However, due to computer limitations, many studies focused on A β fragments, of which A β (16-22) was the most frequently studied.

Solid-state NMR measurements on $A\beta(1-40)$ [100,101], $A\beta(10-35)$ [102,103], $A\beta(16-22)$ [104,105,106], and $A\beta(34-$

42) [76] showed that the first two peptides formed parallel β -sheets whereas the latter two peptides adopted an antiparallel β -sheet organization. Gordon *et al.* showed that amphiphilicity was critical in determining the structural organization of β -sheets in the fibril [106]. Thus, all amyloid fibrils do not share a common supramolecular structure.

A. Assembly of $A\beta(16-22)$

Studies of A β (16-22) assembly are numerous because of the simplicity of the fragment (KLVFFAE), and more importantly, because this fragment comprises the central hydrophobic core (LVFFA) thought to be important in fibril formation of full-length A β . In addition, amino acid substitutions at the E22 position are linked to four familial forms of AD, referred to as Flemish (A21G), Dutch (E22Q), Arctic (E22G), and Italian (E22K). Experimental studies on A β (16-22) showed that A β (16-22) peptides assemble into an antiparallel, in-register, fibrillar structure [104,105,106].

A stability study of $A\beta(16-22)$ octamers packed in different parallel and antiparallel arrangements was conducted by Ma and Nussinov using all-atom MD in explicit solvent and physiological conditions [107]. Antiparallel β -sheet/parallel layer arrangement was found to be the most stable, in agreement with the solid state NMR findings [104]. The same study addressed the stability of $A\beta(16-35)$ and $A\beta(10-35)$. Neither of these two fragments was found to form stable extended β -strands because of the presence of residues G25, S26, N27 with high turn propensities, which create a bend stabilized by a D23-K28 salt bridge. The aim of this study was to understand the supramolecular organization of fibrils, but did not provide any insight into the oligomerization process and possible pathways of assembly.

Klimov and Thirumalai studied formation of an A β (16-22) trimer using all-atom MD in explicit water and a bias to facilitate interactions between peptides [108]. They found that A β (16-22) assembled through multiple pathways, each accompanied by the formation of α -helical intermediates, in agreement with experimental observations [20,21,99]. This tripeptide system, which forms a stable antiparallel β -sheet in water, was shown to be destabilized in urea solution due to hydrogen bond formation between urea and the peptide backbone, which gave rise to stable β -strand monomers [109].

Using the ART and an approximate free energy model, Santini *et al.* showed that an in-register antiparallel β -sheet structure was the most stable structure for the A β (16-22) dimer, despite the existence of several hydrogen bond patterns in both parallel and antiparallel orientations that were thermodynamically possible [110,111]. The existence of alternative β -sheet organizations is important because it helps explain the dependence of β -sheet registry on pH [105] and amino acid composition [112]. Santini *et al.* also found multiple pathways, but α -helical intermediates were not found to be obligatory on the pathway to dimer formation, in contrast to previous findings by Klimov *et al.* [108].

Dimer formation of A β (16-22) and its Arctic (E22G) mutant was studied using all-atom MD with implicit solvent [113]. The results of this study showed that hydrophobic interactions oriented the peptides to minimize the solvent

accessible surface area leading to dimer structures that were kinetically trapped in energetically unfavorable conformations. Once hydrophobic contacts were present, hydrogen bonds formed rapidly in a zipper-like way. A β (16-22) dimers also showed preference for an antiparallel configuration, probably due to the electric dipole moment of this fragment.

An assembly of six $A\beta(16-22)$ peptides was studied by Favrin *et al.* using an unbiased Monte Carlo method with simulated tempering [114]. They used a sequence-based atomic protein model with hydrogen bonding and effective hydrophobic attraction with no explicit water molecules. Favrin *et al.* found different supramolecular structures of $A\beta(16-22)$ aggregates with a high β -strand content. Even though an antiparallel arrangement of β -strands was preferred, parallel arrangement persisted even in the absence of the Coulombic interaction between two charged amino acids, which disagrees with the finding of Klimov *et al.* that Coulombic interactions are the main determinant for the antiparallel orientation in the $A\beta(16-22)$ assembly [108].

Gnanakaran *et al.* applied replica exchange MD simulations at 38 different temperatures in explicit water solvent to study dimer formation of A β (16-22) [115]. They found that a predominant monomer conformation is polyproline-II-like, while there exist six stable dimer conformations, not necessarily limited to only parallel or antiparallel strands. In addition Gnanakaran *et al.* found that water molecules are directly involved in stabilizing certain dimer conformation that cannot be predicted by implicit solvent models.

B. $A\beta(10-35)$ Dimer Stability

Using several methods, Tarus et al. examined the initial steps of A β aggregation by studying the stability of two A β (10-35) dimers in aqueous solution [116]. They generated a pool of possible dimer configurations using a protocol based on shape complementarity. They evaluated these structures using estimates of the desolvation and electrostatic interaction energies to identify putative stable dimer structures. Two globular dimers were identified, the ϕ -dimer that minimizes the desolvation energy of the residues and the ε -dimer that minimizes the interpeptide electrostatic energy. The potential of the mean force associated with the dimerization of the peptides in aqueous solution was computed for both the hydrophobic- and the electrostatic-driven forces using umbrella sampling and all-atom MD simulation at constant temperature and pressure in explicit solvent. This calculation led Tarus et al. to obtain the free energy profiles for each dimer that suggested that the ϕ -dimer has a favorable structure originating from the desolvation of the hydrophobic residues at the interface. Furthermore, they generated MD trajectories originating from the two dimer structures and found that the ϕ -dimer is stabilized primarily through hydrophobic interactions while the ε -dimer is not stable. These results suggested that the preservation of the structure of the central hydrophobic core (L17-A21) plays an important role in the stabilization of the ϕ -dimer structure.

C. Understanding A^β Fibril Assembly

Petkova *et al.* derived a structural model of $A\beta(1-40)$ based on experimental constraints from solid state

NMR [78]. The model is consistent with the cross- β structural motif established by X-ray diffraction [15]. According to this model, the first 10 residues of A β (1-40) are structurally disordered, while segments 12-24 and 30-40 adopt a Bstrand structure. Residues 25-29 contain a backbone bend which brings the two β -strand segments together through side chain-side chain interactions. The supramolecular fibril structure is formed through intermolecular hydrogen bonds along the fibril axis. Petkova et al. suggested that the salt bridge D23-K28 plays a key role in fibril stabilization. The importance of this salt bridge interaction was recently demonstrated by Sciarretta et al. who compared fibrillogenesis of a wild type A β (1-40) with A β (1-40) containing a lactam bridge between the side chains of D23 and K28 (A β (1-40)-Lactam(D23/K28)) [117]. Sciarretta et al. found, in contrast to the wild type, that fibrillogenesis of $A\beta(1-40)$ -Lactam(D23/K28) occurred at a 1000-fold greater rate than observed in A β (1-40) and without a lag phase, possibly bypassing an unfavorable folding step in fibrillogenesis. A β (1-40)-Lactam(D23/K28) also showed a stronger tendency for self-association and formed oligomers at micromolar concentrations, at which the wild type A β (1-40) showed no detectable oligomers.

Intriguingly, in their subsequent work, Petkova *et al.* showed that different $A\beta(1-40)$ fibrillar morphologies have different underlying molecular structures that can be controlled by subtle variations in fibril growth conditions [118]. Moreover, they found that different $A\beta(1-40)$ fibrillar morphologies are associated with different toxicities in neuronal cell cultures.

Recently Lührs et al. reported a three-dimensional structure of $[M35(O)]A\beta(1-42)$ fibrils based on *in vitro* hydrogenbonding constraints from quenched hydrogen/deuteriumexchange NMR, side chain packing constraints from pairwise mutagenesis studies, and parallel, in-register arrangement determined by previous solid-state NMR studies [119]. In $[M35(O)]A\beta(1-42)$, M35 is oxidized, which was shown by several groups to strongly affect A β (1-42) oligomer and fibril formation [120,121,122,123,124]. Results of Lührs et al. showed that the structure of $[M35(O)]A\beta(1-42)$ fibrils consisted of disordered residues 1-17, while residues 18-42 formed a β -strand-turn- β -strand pattern. Denoting the β strand of residues 18-26 by β_1 and the β -strand of residues 31-42 by β_2 , they found that the odd-numbered residues of β_1 , of the *n*th molecule formed contacts with the evennumbered residues of β_2 of the *n*-1th molecule. Between β_1 and β_2 strands within each molecule they found the following important side chain-side chain contacts: D23-K28 (the salt bridge interaction), F19-G38, and A21-V36. Together, these results suggest that $[M35(O)]A\beta(1-42)$ fibrils are stabilized by intermolecular domain swapping-type of side chain interactions.

Using a kinetic theory approach, Massi and Straub derived kinetic equations governing A β deposition and fibril elongation [125]. These equations are applicable to fulllength, fragment, and mutant forms of A β . Their approach is based on the "dock-and-lock" model of A β fibril assembly in which a monomer first "docks" with the fibril end in a reversible manner and then undergoes a conformational reorganization that locks it into place [20,126,127,128,129]. Their kinetic equations consider several possible scenarios for peptide deposition, including fast deposition from solution through an activation/nucleation event and deposition of peptide from solution onto existing fibrils followed by reorganization of the peptide/fibril deposit. The approach of Massi and Straub unifies several views of A β peptide deposition and elongation. They found that the proposed mechanism is consistent with experimental data on the rate of fibril elongation for wild type A β alloforms [130], A β fragments [131,132,81], and mutant A β peptide fragments [129].

A fibrillogenic fragment, $A\beta(12-28)$, was studied by Simona *et al.* using all-atom MD with explicit solvent [133]. Simona *et al.* studied $A\beta(12-28)$ monomer transition from an α -helix to a β -hairpin-like conformation. During this simulation the five consecutive hydrophobic side chains of the central hydrophobic core, LVFFA, were exposed to the solvent, which made the β -hairpin-like conformation prone to aggregation. The results show that while hydrophobic contacts are important to bring together individual molecules in a β hairpin-like conformation, hydrogen bonding and Coulombic interactions are necessary to stabilize the nascent fibrillar aggregates.

Buchete *et al.* applied all-atom MD with explicit solvent and multiple force-fields to probe structural stability of several models of A β (9-40) fibril structure [134]. These structural models were based on solid state NMR data of A β (1-40) by Petkova *et al.* [78]. Buchete *et al.* considered four topologies that were all found to be stable in accord with experimentally-observed structural polymorphism at the molecular level [118]. The inward-pointing salt bridges D23-K28 were found to stabilize the protofilament structure and were hydrated by interior water molecules.

D. Full-Length Aβ Oligomer Formation

Understanding the assembly of full-length A β , especially that of A β (1-40) versus A β (1-42), is critical for gaining insight into oligomerization differences that may be relevant to the particularly strong linkage of the longer A β alloform with disease [135].

Bitan et al. used the method photo-induced cross-linking of unmodified proteins (PICUP) to study oligomerization of full-length A β [136,137]. PICUP enabled identification and quantification of short-lived metastable assemblies with no pre facto structural modification of the native peptide. Bitan *et al.* determined the oligomer size distributions of $A\beta(1-40)$ and A β (1-42) [135]. Whereas the A β (1-40) size distribution comprised primarily monomers through tetramers, the A β (1-42) size distribution was multimodal, displaying peaks in the region of pentamer/hexamer, dodecamer, and octadecamer. The quasi-pentad/hexad periodicity suggested that a basic oligomer building block existed. The term "paranucleus" was used to refer to this pentamer/hexamer unit [135]. Bitan et al. established that I41 of A β (1-42) is essential for formation of paranuclei, while A42 is necessary for assembly of paranuclei into higher-order oligomers. Subsequent studies determined primary structure elements controlling early AB oligomer formation by systematically evaluating 34 physiologically-relevant alloforms [138]. Further studies of Bitan et al. showed that oxidation of M35 blocked A β (1-42) paranucleus formation, producing oligomer distributions indistinguishable from those of A β (1-40) [122].

Recently, Bernstein *et al.* applied mass spectroscopy and ion mobility spectroscopy to the study of $A\beta(1-42)$ and [P19] $A\beta(1-42)$, a substitution that blocks $A\beta$ fibril formation [139]. They observed solution-like structures of $A\beta(1-42)$ that consist of monomers, dimers, tetramers, hexamers, and an aggregate of two hexamers, which correspond to the first step in protofibril formation. These results are consistent with the PICUP findings of Bitan *et al.* [135]. Further experiments by Bernstein *et al.* have extended these findings by demonstrating that monitoring formation of "oligoparanuclear" assemblies is possible using a novel chemical form of $A\beta(1-42)$ and ion mobility spectroscopy detection (Bernstein, unpublished data).

Aggregation propensity of $A\beta(1-40)$ and its fragment $A\beta(1-28)$ was studied by Valerio *et al.* using all-atom MD in an explicit water environment [140]. They used nonlinear signal analysis [141] to study aggregation and folding propensities by considering the distribution of hydrophobic and charged residues. Valerio *et al.* concluded that in the most "aggregation-prone" environment $A\beta(1-40)$ has a higher conformational mobility than does $A\beta(1-28)$ due to the most mobile part of the molecule, the hydrophobic tail, which is not present in $A\beta(1-28)$.

The first DMD study of $A\beta(1-40)$ aggregation was done by Peng *et al.* who used a two-bead peptide model with $G\bar{o}$ interactions based on the $A\beta(1-40)$ structure in a membranelike environment [142]. Peng *et al.* showed that molecules assemble into fibril-like aggregates with parallel, in-register organization.

Folding and dimer formation of $A\beta(1-40)$ and $A\beta(1-42)$ was studied by Urbanc et al. using a combination of DMD and all-atom MD simulations [143]. The explicit solvent/implicit solvent MD method was applied to estimate the free energy of different dimer conformations of both $A\beta(1-$ 40) and A β (1-42). DMD simulations using a four-bead protein model with hydrogen bond interactions [70] predicted a folded A β (1-42) monomer with a β -hairpin structure and a turn at residues G25-S26-N27. Two such β-hairpin monomers assembled into planar β -sheet dimer conformations, which then were studied for stability using all-atom MD. Urbanc *et al.* showed that all planar β -sheet dimers had a higher free energy than the corresponding monomeric states and that there was no significant free energy difference between A β (1-40) and A β (1-42) dimers. These results suggest that dimer conformations other than planar β -sheets are responsible for the experimentally-observed differences in oligomerization between A β (1-40) and of A β (1-42) [135]. At the molecular level, this result implies that interactions other than hydrogen bond interactions are needed to study A β oligomer formation in DMD.

Oligomer formation of A β (1-40) and A β (1-42) was studied by Urbanc *et al.* using DMD and the four-bead protein model with hydrogen bonding and amino acid-specific interactions [94]. Initially, the separated A β peptides folded into collapsed coil structures and then assembled into oligomers of different sizes [94]. After these simulations reached a steady state and the oligomer size distribution did not significantly change in time, the respective $A\beta(1-40)$ and $A\beta(1-42)$ size distributions differed significantly. $A\beta(1-42)$ formed significantly more pentamers than $A\beta(1-40)$ and $A\beta(1-40)$ formed significantly more dimers than did $A\beta(1-42)$. These results showed that the effective hydrophobic attraction of I41 has a direct impact on formation of $A\beta(1-42)$ oligomers. Urbanc *et al.* also pointed out structural differences between $A\beta(1-40)$ and $A\beta(1-42)$ oligomers that are amenable to experimental study.

V. DISCUSSION

In a recent review, Snow et al. pointed out that in recent years important advances in validation of computer simulation methodology have been made possible through quantitative comparisons with experiments for small, rapidly-folding proteins [144]. These advances also have influenced computational studies of amyloidogenic proteins, in particular A β . In this review, we have discussed the unique insights into $A\beta$ folding and assembly into oligomers and fibrils that can be obtained through computer simulation methods. Along with the simulation results, we also have reviewed experimental findings to emphasize the importance of bidirectional in vi*tro* interactions for obtaining biologically-relevant information on A β structure and dynamics. In contrast to understanding fibril structure, for which a combination of NMR methods and pairwise mutagenesis recently yielded a $[M35(O)]A\beta(1-42)$ fibrillar structure with atomic-level details [119], experiments seeking to elucidate the structure of early oligomers are significantly more challenging. Computational studies of full-length A β fibrils using all-atom MD currently are limited primarily to the investigation of the stability of pre-constructed AB fibril models [145]. In contrast, the elucidation of early folding events and oligomerization is more computationally accessible and can potentially yield new information amenable to in vitro testing.

A hierarchy of computational approaches extending from the most coarse-grained models with simplified dynamics, through intermediate resolution models such as four-bead and united-atom, to all-atom models with MD in explicit solvent is needed to understand the structure and dynamics of A β at the molecular level. Different levels of simplicity in the model and dynamics can address different questions. Allatom MD with explicit solvent is the most realistic approach and the only one that can address questions related to solvation of proteins, but it cannot be applied to study processes that occur in a time regime exceeding ~100-200 ns. The most coarse-grained approaches, e.g., one-, two-, or fourbead DMD models with implicit solvent, are efficient enough to study oligomer or fibril formation starting from initially-separated peptides. However, the degree of detail strongly depends on the protein model and implemented amino acid-specific interactions.

There are differences among the coarse-grained protein models in which an amino acid is represented by one, two, four, six, or all heavy atom beads. Although one- and twobead models are too coarse-grained to model realistically the protein backbone, they allow the study of general properties of protein folding in more complex environments. An example of such an application is a recent study of molecular crowding effects on native state stability and protein folding

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dynamics [146]. The models with four or more beads per amino acid, i.e., intermediate resolution models, have a welldescribed backbone geometry. If, in addition, the backbone hydrogen bond is implemented, these intermediate resolution models are capable of accounting for α -helix and β -hairpin monomer conformations by simply varying the temperature, and most importantly, without any knowledge of the native protein state [70]. Using a four-bead model of polyalanine, Nguyen and Hall studied mechanisms of fibril formation [147,148,149]. They derived a phase diagram in dependence on temperature and molar concentration that shows regions of amorphous aggregates, fibrillar aggregates, as well as regions with no aggregation [147,148,149]. Ding et al. recently used a coarse-grained model with four backbone beads and up to several side chain beads to fold a small Trpcage protein to an NMR-resolution native structure, demonstrating that all-atom protein description is not necessary to successfully simulate folding into a native state [71]. All these properties of intermediate resolution models, and the above validation studies, illustrate that these models can form a solid foundation upon which ab initio modeling of specific proteins can be conducted.

Experimental findings motivate the development of computational approaches. Can computer simulation findings direct in vitro research? Results of DMD simulations of $A\beta(1-40)$ and $A\beta(1-42)$ oligomer formation suggest the answer is "yes" [94]. These results are consistent with existing experimental data, which defined the roles of individual amino acids in oligomerization of both alloforms [122,135,138]. More importantly, they yield new structural predictions. Even though the main difference in conformational dynamics between the two alloforms was driven by the hydrophobic residues I41 and A42, the simulations showed that the N-termini of A β (1-40) and A β (1-42) oligomers were arrayed in a structurally distinct manner. The 3D structure of oligomers is tightly related to their neurotoxic properties, thus the N-terminal structural differences between AB alloforms revealed through simulation provide a theoretical basis for direct experimental testing of structure-neurotoxicity correlations in vitro and in vivo.

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ABBREVIATIONS

- $A\beta$ = Amyloid β -protein
- APP = Amyloid β -protein precursor
- MD = Molecular dynamics
- DMD = Discrete molecular dynamics
- ART = Activation-relaxation technique

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